A C-GLUCOSIDE OF CHAMAZULENE

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For the purposes of studying the biological activity of natural azulenes and their synthetic analogs it is of great interest to obtain water-soluble azulene compounds.

Work in this direction has been carried out only by Treibs, and this has been concerned with the synthesis of Cglucosides of the azulenes [1, 2]. When azulene, 1-isopropyl-5-methylazulene, and guaiazulene were heated with aacetobromoglucose, he obtained the corresponding C-glucosides.

Treibs also used guaiazulene to obtain the C-glucosides of some other mono-, di- and polysaccharides. The C_1 or the C_3 of the five-membered ring of the azulenes took part in the C-glucosidation reaction. Better results were achieved in the synthesis of the C-glucosides of guaiazulene. The derivatives that Treibs synthesized retained their biological activity [2].

We have attempted to obtain a C-glucoside of natural chamazulene. Using the method recommended by Treibs, with some modifications, we have succeeded in isolating from the reaction mixture a water-soluble acetate of C-glucoside of chamazulene in the form of blue crystals with mp 183° C.

Found, %: C 65. 43; H 6. 64. Calculated for C28H34O9, %: C 65. 33; H 6. 66.

By saponifying the acetate of the glucoside with sodium ethoxide, we obtained the C-glucoside of chamazulene as light blue crystals with mp $142^{\circ}-146^{\circ}$ C.

Found, %: C 68.48; H 7.76. Calculated for C₂₀H₂₆O₅, %: C 69.32; H 7.57.

The substance dissolved readily in water, forming a blue solution.

REFERENCES

1. W. Treibs, Ann. 667, 141, 1963.

2. W. Treibs, W. German patent no. 1 179 933, 22 October 1964.

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